

2023-2024

# High School ISAT Science Blueprint



IDAHO STATE DEPARTMENT OF EDUCATION  
ASSESSMENT | ISAT SCIENCE

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## INTRODUCTION

This document is comprised of three different sections.

1. **Operational Items by Domain**: A table containing the number of Operational items for each domain on the Elementary Science ISAT.
2. **Test Item Breakdown**: A table showing the operational and field test item counts on the test. In addition to the 19 operational items on the live test, every student is assigned either one cluster or a few stand-alone items that are being field tested. Please note that field test items do not count towards a student’s score and always appear at the end of the assessment.
3. **Potential Standards Addressed**: A list of the possible standards that can be used within each domain. [The 2022 Idaho State Science Standards](#) will be assessed during the 2023-2024 school year.

### 1. OPERATIONAL ITEMS BY DOMAIN

Domain	# of Clusters	# of Stand-Alone Items	Total Domain Items
Earth and Space Science	2	4	6
Life Science	2	4	6
Physical Science	2	5	7

### 2. TEST ITEM BREAKDOWN

Item Type	# of Clusters	# of Stand-Alone Items	Total Items
Operational Items	6	13	19
Field Test Items	1 or	up to 4	1 or up to 4
Total Summative Test Length	7 or	up to 17	20 or up to 23

### 3. POTENTIAL STANDARDS ASSESSED

#### Earth and Space Science

##### Earth's Place in the Universe

Idaho State Science Standard	Idaho State Science Standard Text
HS-ESS-1.1	Develop a model based on evidence to illustrate the life span of the Sun and the role of nuclear fusion in the Sun's core to release energy that eventually reaches Earth in the form of radiation.
HS-ESS-1.2	Construct an explanation of the current model of the origin of the universe based on astronomical evidence of light spectra, motion of distant galaxies, and composition of matter in the universe.
HS-ESS-1.3	Communicate scientific ideas about the way stars, over their life cycle, produce elements.
HS-ESS-1.4	Use mathematical or computational representations to predict the motion of orbiting objects in the solar system.
HS-ESS-1.5	Evaluate evidence of the past and current movements of continental and oceanic crust and the theory of plate tectonics to explain the ages of crustal rocks.
HS-ESS-1.6	Apply scientific reasoning and evidence from ancient Earth materials, meteorites, and other planetary surfaces to construct an account of Earth's formation and early history.

##### Earth's Systems

Idaho State Science Standard	Idaho State Science Standard Text
HS-ESS-2.1	Develop a model to illustrate how Earth's internal and surface processes operate at different spatial and temporal scales to form continental and ocean-floor features.

<b>Idaho State Science Standard</b>	<b>Idaho State Science Standard Text</b>
HS-ESS-2.2	Analyze geoscience data to make the claim that one change to Earth’s surface can create feedbacks that cause changes to other Earth systems.
HS-ESS-2.3	Develop a model based on evidence of Earth’s interior to describe the cycling of matter by thermal convection.
HS-ESS-2.4	Use a model to describe how variations in the flow of energy into and out of Earth’s systems result in changes in climate.
HS-ESS-2.5	Plan and conduct an investigation of the properties of water and its effects on Earth materials and surface processes.
HS-ESS-2.6	Develop a model to describe the cycling of carbon among the hydrosphere, atmosphere, geosphere, and biosphere.
HS-ESS-2.7	Construct an argument based on evidence about the simultaneous coevolution of Earth’s systems and life on Earth.

#### **Earth and Human Activity**

<b>Idaho State Science Standard</b>	<b>Idaho State Science Standard Text</b>
HS-ESS-3.1	Construct an explanation based on evidence for how the availability of natural resources, occurrence of natural hazards, and changes in climate have influenced human activity.
HS-ESS-3.2	Evaluate competing design solutions for developing, managing, and utilizing energy and mineral resources based on cost-benefit ratios.
HS-ESS-3.3	Illustrate relationships among management of natural resources, the sustainability of human populations, and biodiversity.

Idaho State Science Standard	Idaho State Science Standard Text
HS-ESS-3.4	Evaluate or refine a scientific or technological solution that mitigates or enhances human influences on natural systems.
HS-ESS-3.5	Analyze geoscience data and the results from global climate models to make an evidence-based explanation of how climate variability can affect Earth's systems on a global and regional scale.
HS-ESS-3.6	Communicate how relationships among Earth systems are being influenced by human activity

## Life Science

### From Molecules to Organisms: Structure and Process

Idaho State Science Standard	Idaho State Science Standard Text
HS-LS-1.1	Construct an explanation based on evidence for how the structure of DNA determines the structure of proteins which carry out the essential functions of life through systems of specialized cells.
HS-LS-1.2	Develop and use a model to illustrate the hierarchical organization of interacting systems that provide specific functions within multicellular organisms.
HS-LS-1.3	Plan and conduct an investigation to provide evidence that feedback mechanisms maintain homeostasis.
HS-LS-1.4	Use a model to illustrate the role of cellular division (mitosis) and differentiation in producing and maintaining complex organisms.
HS-LS-1.5	Use a model to illustrate how photosynthesis transforms light energy into stored chemical energy.
HS-LS-1.6	Construct and revise an explanation based on evidence for how carbon, hydrogen, and oxygen from sugar molecules may combine with other elements to form amino acids and/or other large carbon-based molecules.

Idaho State Science Standard	Idaho State Science Standard Text
HS-LS-1.7	Use a model to illustrate that cellular respiration is a chemical process whereby the bonds of food molecules and oxygen molecules are broken and the bonds in new compounds are formed resulting in a net transfer of energy.

**Ecosystems: Interactions, Energy, and Dynamics**

Idaho State Science Standard	Idaho State Science Standard Text
HS-LS-2.1	Use mathematical and/or computational representations to support explanations of factors that affect carrying capacity of ecosystems at different scales.
HS-LS-2.2	Use mathematical representations to support explanations that biotic and abiotic factors affect biodiversity at different scales within an ecosystem.
HS-LS-2.3	Construct an explanation using mathematical representations to support claims for the flow of energy through trophic levels and the cycling of matter in an ecosystem.
HS-LS-2.4	Develop a model to illustrate the role of photosynthesis and cellular respiration in the cycling of carbon among the biosphere, atmosphere, hydrosphere, and geosphere.
HS-LS-2.5	Evaluate the claims, evidence, and reasoning that changing the conditions of a static ecosystem may result in a new ecosystem.
HS-LS-2.6	Design, evaluate, and/or refine practices used to manage a natural resource based on direct and indirect influences of human activities on biodiversity and ecosystem health
HS-LS-2.7	Evaluate the evidence for the role of group behavior on individual and species' ability to survive and reproduce.

### Heredity: Inheritance and Variation of Traits

Idaho State Science Standard	Idaho State Science Standard Text
HS-LS-3.1	Ask questions to clarify relationships about the role of DNA and chromosomes in coding the instructions for characteristic traits passed from parents to offspring.
HS-LS-3.2	Make and defend a claim based on evidence that inheritable genetic variations may result from: (1) new genetic combinations through meiosis, (2) viable errors occurring during replication, and/or (3) mutations caused by environmental factors.
HS-LS-3.3	Apply concepts of statistics and probability to explain the variation and distribution of expressed traits in a population.

### Biological Adaptation: Unity and Diversity

Idaho State Science Standard	Idaho State Science Standard Text
HS-LS-4.1	Communicate scientific information that common ancestry and biological evolution are supported by multiple lines of empirical evidence.
HS-LS-4.2	Construct an explanation based on evidence that the process of evolution, through the mechanism of natural selection, primarily results from four factors: (1) the potential for a species to increase in number, (2) the heritable genetic variation of individuals in a species due to mutation and sexual reproduction, (3) competition for limited resources, and (4) the proliferation of those organisms that are better able to survive and reproduce in the environment.



Idaho State Science Standard	Idaho State Science Standard Text
HS-LS-4.3	Apply concepts of statistics and probability to support explanations that organisms with an advantageous heritable trait tend to increase in proportion to organisms lacking this trait.
HS-LS-4.4	Construct an explanation based on evidence for how natural selection leads to adaptation of populations.
HS-LS-4.5	Evaluate models that demonstrate how changes in an environment may result in the evolution of a population of a given species; the emergence of new species over generations; or the extinction of other species due to the processes of genetic drift, gene flow, mutation, and natural selection.

## Physical Science

### Structure and Properties of Matter and Chemical Reactions (Chemistry)

Idaho State Science Standard	Idaho State Science Standard Text
HS-PSC-1.1	Develop models to describe the atomic composition of simple molecules and extended structures.
HS-PSC-1.2	Use the periodic table as a model to predict the relative properties of elements based on the patterns of electrons in the outermost energy level of atoms.
HS-PSC-1.3	Plan and conduct an investigation to gather evidence to compare the structure of substances at the bulk scale to infer the strength of electrostatic forces between particles.
HS-PSC-1.4	Develop models to illustrate the changes in the composition of the nucleus of the atom and the energy released during the processes of fission, fusion, and other types of radioactive decay.
HS-PSC-1.5	Communicate scientific and technical information about why the molecular-level structure is important in the functioning of designed materials.

<b>Idaho State Science Standard</b>	<b>Idaho State Science Standard Text</b>
HS-PSC-2.1	Construct and revise an explanation for the outcome of a simple chemical reaction based on the outermost electron states of atoms, trends in the periodic table, and knowledge of the patterns of chemical properties.
HS-PSC-2.2	Develop a model to illustrate that the energy transferred during an exothermic or endothermic chemical reaction is based on the bond energy difference between bonds broken (absorption of energy) and bonds formed (release of energy).
HS-PSC-2.3	Apply scientific principles and evidence to provide an explanation about the effects of changing the temperature or concentration of the reacting particles on the rate at which a reaction occurs.
HS-PSC-2.4	Use mathematical representations to support the claim that the number and type of atoms, and therefore mass, are conserved during a chemical reaction.

### **Energy (Chemistry)**

<b>Idaho State Science Standard</b>	<b>Idaho State Science Standard Text</b>
HS-PSC-3.1	Ask questions to clarify the idea that electromagnetic radiation can be described either by a wave model or a particle model.
HS-PSC-3.2	Create a computational model to calculate the change in the energy of one component in a system when the change in energy of the other component(s) and energy flows in and out of the system are known.

Idaho State Science Standard	Idaho State Science Standard Text
HS-PSC-3.3	Develop and use models to illustrate that energy at the macroscopic scale can be accounted for as a combination of energy associated with the motions of particles (objects) and energy associated with the relative positions of particles (objects).
HS-PSC-3.4**OPTIONAL	Design, build, and refine a device that works within given constraints to convert one form of energy into another form of energy.
HS-PSC-3.5	Plan and conduct an investigation to provide evidence that the transfer of thermal energy when two components of different temperature are combined within a closed system results in a more uniform energy distribution among the components in the system (second law of thermodynamics).

#### Motion and Stability: Forces and Interactions (Physics)

Idaho State Science Standard	Idaho State Science Standard Text
HS-PSP-1.1	Analyze data to support the claim that Newton’s second law of motion describes the mathematical relationship among the net force on a macroscopic object, its mass, and its acceleration.
HS-PSP-1.2	Use mathematical representations to support the claim that the total momentum of a system of objects is conserved when there is no net force on the system.
HS-PSP-1.3	Apply scientific and engineering ideas to design, evaluate, and refine a device that minimizes the force on a macroscopic object during a collision.
HS-PSP-1.4	Use mathematical representations of Newton’s Law of Gravitation and Coulomb’s Law to describe and predict the gravitational and electrostatic forces between objects.

Idaho State Science Standard	Idaho State Science Standard Text
HS-PSP-1.5	Plan and conduct an investigation to provide evidence that an electric current can produce a magnetic field and that a changing magnetic field can produce an electric current.
HS-PSP-1.6	Communicate scientific and technical information about why the molecular-level structure is important in the functioning of designed materials.

### Energy (Physics)

Idaho State Science Standard	Idaho State Science Standard Text
HS-PSP-2.1	Create a computational model to calculate the change in the energy of one component in a system when the change in energy of the other component(s) and energy flows in and out of the system are known.
HS-PSP-2.2	Develop and use models to illustrate that energy at the macroscopic scale can be accounted for as a combination of energy associated with the motions of particles (objects) and energy associated with the relative positions of particles (objects).
HS-PSP-2.3	Design, build, and refine a device that works within given constraints to convert one form of energy into another form of energy.
HS-PSP-2.4	Plan and conduct an investigation to provide evidence that the transfer of thermal energy when two components of different temperature are combined within a closed system results in a more uniform energy distribution among the components in the system (second law of thermodynamics).
HS-PSP-2.5	Develop and use a model of two objects interacting through electric or magnetic fields to illustrate the forces between objects and the changes in energy of the objects due to the interaction.

### Waves (Physics)

Idaho State Science Standard	Idaho State Science Standard Text
HS-PSP-3.1	Use mathematical representations to support a claim regarding relationships among the frequency, wavelength, and speed of waves traveling in various media.
HS-PSP-3.2	Evaluate questions about the advantages of using digital transmission and storage of information.
HS-PSP-3.3	Evaluate the claims, evidence, and reasoning behind the idea that electromagnetic radiation can be described either by a wave model or a particle model, and that for some situations one model is more useful than the other.
HS-PSP-3.4	Evaluate the validity and reliability of claims in published materials of the effects that different frequencies of electromagnetic radiation have when absorbed by matter.
HS-PSP-3.5	Communicate technical information about how some technological devices use the principles of wave behavior and wave interactions with matter to transmit and capture information and energy.